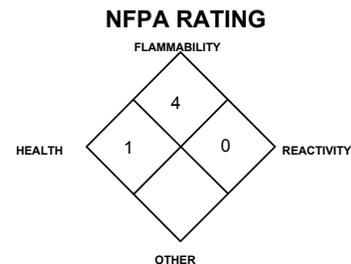




# MATERIAL SAFETY DATA SHEET

Prepared to U.S. OSHA, CMA, ANSI and Canadian WHMIS Standards



## **PART I** *What is the material and what do I need to know in an emergency?*

### 1. PRODUCT IDENTIFICATION

CHEMICAL NAME; CLASS:

**FLAMMABLE GAS MIXTURE  
LPG LIQUID STANDARD**

September 23, 2009

MIXTURE DESCRIPTION:

Flammable, liquefied gas

PRODUCT NAME :

HCR188C (R441A) Hydrocarbon Refrigerant (EcoCool)

SUPPLIER/MANUFACTURER'S NAME:

ComStar International Inc.

ADDRESS:

20-45 128 Street  
College Point, NY 11356

BUSINESS PHONE:

1-800-328-0142

### 2. COMPOSITION and INFORMATION ON INGREDIENTS

CHEMICAL NAME	CAS #	mole %	EXPOSURE LIMITS IN AIR					
			ACGIH		OSHA			OTHER ppm
			TLV ppm	STEL ppm	PEL ppm	STEL ppm	IDLH ppm	
Propane	74-98-6	0.1-80	Simple Asphyxiant		1000	NE	2100	NIOSH REL: 1000 DFG MAK: 1000
n-Butane	106-97-8	0.1-50	800	NE	800 (Vacated 1989 PEL)	NE	NE	NIOSH REL: 800 DFG MAK: 1000
Isobutane	75-28-5	0.1-50	There are no specific exposure limits for Isobutane. Isobutane is a simple asphyxiant (SA). Oxygen levels should be maintained above 19.5%.					
Ethane	74-84-0	0.1-25	Ethane is a simple asphyxiant (SA).					

NE = Not Established

See Section 16 for Definitions of Terms Used.

NOTE: All WHMIS required information is included. It is located in appropriate sections based on the ANSI Z400.1-1993 format.

### 3. HAZARD IDENTIFICATION

**EMERGENCY OVERVIEW:** This product is a colorless, liquefied, flammable gas with an ether-like odor. This product poses a serious fire hazard when accidentally released. The gas may spread long distances, distant flashback and ignition are possible. This gas mixture can also cause symptoms of oxygen deprivation (i.e. dizziness, breathing rate increase). Contact with rapidly expanding gases may cause frostbite. Emergency responders must wear personal protective equipment appropriate for the situation to which they are responding.

**SYMPTOMS OF OVER-EXPOSURE BY ROUTE OF EXPOSURE:** The most significant route of over-exposure for this product is by inhalation.

**INHALATION:** At high concentrations Propane, a component of this product, can act as a narcotic. This product can cause symptoms of oxygen deprivation (asphyxiation) when presented in high enough concentrations to significantly lower oxygen concentration. Individuals breathing such an atmosphere may experience symptoms which include headaches, ringing in ears, dizziness, drowsiness, unconsciousness, nausea, vomiting, and depression of all the senses. Under some circumstances of over-exposure, death may occur. It should be noted that before suffocation could occur, the lower flammability limit of propane in air would be exceeded; possibly causing an oxygen-deficient and explosive atmosphere. The effects associated with various levels of oxygen are as follows:

12-16% Oxygen: Breathing and pulse rate increased, muscular coordination slightly disturbed.  
 10-14% Oxygen: Emotional upset, abnormal fatigue, disturbed respiration.  
 6-10% Oxygen: Nausea and vomiting, collapse or loss of consciousness.  
 Below 6%: Convulsive movements, possible respiratory collapse, and death.

**OTHER POTENTIAL HEALTH EFFECTS:** Contact with rapidly expanding gases (which are released under high pressure) may cause frostbite. Symptoms of frostbite include change in skin color to white or grayish-yellow. The pain after such contact can quickly subside.

**HEALTH EFFECTS OR RISKS FROM EXPOSURE:** An Explanation in Lay Terms. Over-exposure to this product may cause the following health effects:

**ACUTE:** This gas mixture can cause symptoms of oxygen deprivation and presents a health hazard by displacing the oxygen in the atmosphere. Contact with rapidly expanding gases (which are released under high pressure) may cause frostbite. Symptoms of frostbite include change in skin color to white or grayish-yellow.

**CHRONIC:** There are currently no known adverse health effects associated with chronic exposures to the components of this compressed gas.

HAZARDOUS MATERIAL INFORMATION SYSTEM			
<b>HEALTH</b>	(BLUE)	1	
<b>FLAMMABILITY</b>	(RED)	4	
<b>REACTIVITY</b>	(YELLOW)	0	
<b>PROTECTIVE EQUIPMENT</b>			<b>B</b>
EYES	RESPIRATORY	HANDS	BODY
	See Section 8		See Section 8
For routine industrial applications			

## PART II *What should I do if a hazardous situation occurs?*

### 4. FIRST-AID MEASURES

**RESCUERS SHOULD NOT ATTEMPT TO RETRIEVE VICTIMS OF EXPOSURE TO THIS PRODUCT WITHOUT ADEQUATE PERSONAL PROTECTIVE EQUIPMENT. At a minimum, Self-Contained Breathing Apparatus Personal Protective equipment should be worn.**

Remove victim(s) to fresh air, as quickly as possible. In case of eye contact which leads to irritation, immediately flush eyes with copious amounts of water for at least 15 minutes. If not breathing, give artificial respiration. If breathing is difficult, give oxygen. Only trained personnel should administer supplemental oxygen.

In case of frostbite, place the frostbitten part in warm water. **DO NOT USE HOT WATER.** If warm water is not available, or is impractical to use, wrap the affected parts gently in blankets. Alternatively, if the fingers or hands are frostbitten, place the affected area in the armpit. Encourage victim to gently exercise the affected part while being warmed. Seek immediate medical attention.

## 4. FIRST-AID MEASURES (Continued)

Victim(s) must be taken for medical attention. Rescuers should be taken for medical attention, if necessary. Take copy of label and MSDS to physician or other health professional with victim(s).

## 5. FIRE-FIGHTING MEASURES

**FLASH POINT, (method):** Not determined for this product. For Propane, the main component, the flashpoint is: -104 °C; -156 °F (Closed Cup)

**AUTOIGNITION TEMPERATURE:** 450 °C; 842 °F (for Propane)

**FLAMMABLE LIMITS (in air by volume, %):**

Lower (LEL): 2.2 for Propane

Upper (UEL): 9.5% for Propane

**FIRE EXTINGUISHING MATERIALS:** Extinguish fire by shutting-off the source of the gas. Use water spray to cool fire-exposed containers, structures, and equipment.

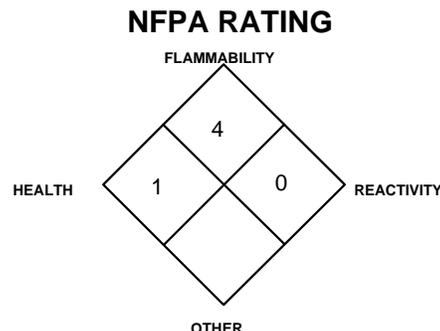
**UNUSUAL FIRE AND EXPLOSION HAZARDS:** When involved in a fire, this material may decompose and produce toxic gases including carbon monoxide and carbon dioxide. Additionally, when involved in fire, the cylinders may rupture. Vapors of this product can travel long distances to an ignition source and flashback.

**Explosion Sensitivity to Mechanical Impact:** Not Sensitive.

**Explosion Sensitivity to Static Discharge:** Static discharge may cause this product to ignite explosively.

**DANGER!** Fires impinging (direct flame) on the outside surface of unprotected pressure storage vessels of this product can be very dangerous. The resulting fire and explosion can result in severe equipment damage and personnel injury or death over a large area around the vessel. For massive fires in large areas, use unmanned hose holder or monitor nozzles; if this is not possible, withdraw from area and allow fire to burn.

**SPECIAL FIRE-FIGHTING PROCEDURES:** Structural fire-fighters must wear Self-Contained Breathing Apparatus and full protective equipment. The best fire-fighting technique may be simply to let the burning gas escape from the pressurized cylinder, tank car, or pipeline. Stop the leak before extinguishing fire. If the fire is extinguished before the leak is sealed, the still-leaking gas could explosively re-ignite without warning and cause extensive damage, injury, or fatality. In this case, increase ventilation (in enclosed areas) to prevent flammable or explosive mixture formation. Evacuation may be necessary. Refer to the North American Emergency Response Guidebook for additional information.



## 6. ACCIDENTAL RELEASE MEASURES

**SPILL AND LEAK RESPONSE:** Uncontrolled releases should be responded to by trained personnel using pre-planned procedures. Proper protective equipment should be used. In case of a release, clear the affected area, protect people, and respond with trained personnel. Adequate fire protection must be provided.

Minimum Personal Protective Equipment should be **Level B: Fire retardant protective clothing, mechanically-resistant gloves and Self-Contained Breathing Apparatus.** Use only non-sparking tools and equipment. Locate and seal the source of the leaking gas. Protect personnel attempting the shut-off with water-spray. Allow the gas to dissipate. Monitor the surrounding area for Oxygen and combustible gas levels. The atmosphere must have at least 19.5 percent oxygen before personnel can be allowed in the area without Self-Contained Breathing Apparatus. Combustible gas concentrations must be below 10% of the LEL for Propane (LEL = 2.2) prior to entry. Attempt to close the main source valve prior to entering the area. If this does not stop the release (or if it is not possible to reach the valve), allow the gas to release in-place or remove it to a safe area and allow the gas to be released there.

## PART III *How can I prevent hazardous situations from occurring?*

### 7. HANDLING and STORAGE

**WORK PRACTICES AND HYGIENE PRACTICES:** As with all chemicals, avoid getting this product IN YOU. Do not eat or drink while handling chemicals. Be aware of any signs of dizziness or fatigue; exposures to fatal concentrations of this product could occur without any significant warning symptoms. All work operations should be monitored in such a way that emergency personnel can be immediately contacted in the event of a release. Non sparking tools should be used.

**STORAGE AND HANDLING PRACTICES:** Cylinders should be stored in dry, well-ventilated areas away from sources of heat. Compressed gases can present significant safety hazards. Store containers away from heavily trafficked areas and emergency exits. Store cylinders of this gas mixture away from incompatible materials.

**SPECIAL PRECAUTIONS FOR HANDLING GAS CYLINDERS:** Protect cylinders against physical damage. Store in cool, dry, well-ventilated area, away from sources of heat, ignition and direct sunlight. Do not allow area where cylinders are stored to exceed 51 °C (125 °F). Use a check valve or trap in the discharge line to prevent hazardous backflow. Cylinders should be stored upright and be firmly secured to prevent falling or being knocked over. Cylinders can be stored in the open, but in such cases, should be protected against extremes of weather and from the dampness of the ground to prevent rusting. Never tamper with pressure relief devices in valves and cylinders. The following rules are applicable to situations in which cylinders are being used :

**Before Use:** Move cylinders with a suitable hand-truck. Do not drag, slide or roll cylinders. Do not drop cylinders or permit them to strike each other. Secure cylinders firmly. Leave the valve protection cap in-place until cylinder is ready for use.

**During Use:** Use designated CGA fittings and other support equipment. Do not use adapters. Do not heat cylinder by any means to increase the discharge rate of the product from the cylinder. Use check valve or trap in discharge line to prevent hazardous backflow into the cylinder. Do not use oils or grease on gas-handling fittings or equipment.

**After Use:** Close main cylinder valve. Replace valve protection cap. Mark empty cylinders "EMPTY".

**NOTE:** Use only DOT or ASME code containers. Earth-ground and bond all lines and equipment associated with this product. Close valve after each use and when empty. Cylinders must not be recharged except by or with the consent of owner. For additional information refer to the Compressed Gas Association Pamphlet P-1, *Safe Handling of Compressed Gases in Containers*. Additionally, refer to CGA Bulletin SB-2 "Oxygen Deficient Atmospheres".

**PROTECTIVE PRACTICES DURING MAINTENANCE OF CONTAMINATED EQUIPMENT:** Follow practices indicated in Section 6 (Accidental Release Measures). Make certain application equipment is locked and tagged-out safely. Purge gas handling equipment with inert gas (i.e. nitrogen) before attempting repairs. Always use product in areas where adequate ventilation is provided.

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## 8. EXPOSURE CONTROLS - PERSONAL PROTECTION

**VENTILATION AND ENGINEERING CONTROLS:** Use with adequate ventilation to ensure exposure limits are maintained below those described in Section 2 (Composition and Information on Ingredients). Local exhaust ventilation is preferred, because it prevents dispersion of this gas mixture into the work place by eliminating it at its source. If appropriate, install automatic monitoring equipment to detect the level of Oxygen. Eye wash stations/safety showers should be near areas where this product is used or stored.

**RESPIRATORY PROTECTION:** Maintain oxygen levels above 19.5% in the workplace. Use supplied air respiratory protection if oxygen levels are below 19.5% or during emergency response to a release of this product. If respiratory protection is required, follow the requirements of the Federal OSHA Respiratory Protection Standard (29 CFR 1910.134), or equivalent State standards. For additional information, the NIOSH recommended protection guidelines for Propane are provided as follows:

NIOSH RECOMMENDATIONS FOR PROPANE CONCENTRATIONS IN AIR:

UP TO 2100 PPM: Supplied-air respirator operated or Self Contained Breathing Apparatus with a full facepiece.

EMERGENCY OR PLANNED ENTRY INTO UNKNOWN CONCENTRATIONS OR IDLH CONDITIONS:

Positive pressure, full-facepiece Self Contained Breathing Apparatus; or positive pressure, full-facepiece Self Contained Breathing Apparatus with an auxiliary positive pressure Self Contained Breathing Apparatus.

ESCAPE:

Escape-type Self Contained Breathing Apparatus.

**EYE PROTECTION:** Splash goggles, face-shields or safety glasses.

**HAND PROTECTION:** Wear mechanically-resistant gloves when handling cylinders of this product. Wear chemically-resistant gloves when using this gas mixture. Butyl rubber, chlorinated polyethylene, neoprene nitrile, and polyvinyl rubber are recommended.

**BODY PROTECTION:** Use body protection appropriate for task.

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## 9. PHYSICAL and CHEMICAL PROPERTIES

The following information is for Propane, the main component of this product:

**VAPOR DENSITY:** 1.868 kg/m<sup>3</sup>

**SPECIFIC GRAVITY (air =1):** 1.550

**SOLUBILITY IN WATER:** Very slight.

**EVAPORATION RATE (nBuAc = 1):** Not applicable.

**ODOR THRESHOLD:** 22,000-36,000 mg/m<sup>3</sup>

**COEFFICIENT WATER/OIL DISTRIBUTION:** 2.36

**APPEARANCE AND COLOR:** This product is a colorless gas with a natural gas smell.

**HOW TO DETECT THIS SUBSTANCE (warning properties):** There are no distinct warning properties. In terms of leak detection, fittings and joints can be painted with a soap solution to detect leaks, which will be indicated by a bubble formation.

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**pH:** Not applicable.

**FREEZING POINT:** -187.70 °C (-305.9°F)

**BOILING POINT(°F @ 1 atm.):** (-43.7 °F) -42 °C

**EXPANSION RATIO:** Not available.

**VAPOR PRESSURE (psia):** 123.7

**SPECIFIC VOLUME (ft<sup>3</sup>/lb):** 8.7

## 10. STABILITY and REACTIVITY

**STABILITY:** Normally stable in gaseous state.

**DECOMPOSITION PRODUCTS:** When ignited in the presence of oxygen, this gas will burn to produce carbon monoxide and carbon dioxide.

**MATERIALS WITH WHICH SUBSTANCE IS INCOMPATIBLE:** This product is not compatible with strong oxidizers.

**HAZARDOUS POLYMERIZATION:** Will not occur.

**CONDITIONS TO AVOID:** Contact with incompatible materials and exposure to heat, sparks and other sources of ignition. Cylinders exposed to high temperatures or direct flame can rupture or burst.

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## **PART III** *How can I prevent hazardous situations from occurring?*

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### **11. TOXICOLOGICAL INFORMATION**

**TOXICITY DATA:** The following toxicology data are for the components of this gas mixture present at a level greater than 1 mole %:

**N-BUTANE:**

LC<sub>50</sub> (inhalation-rat) = 658,000 mg/m<sup>3</sup>  
LC<sub>50</sub> (inhalation-mouse) = 680,000 mg/m<sup>3</sup>

**ISOBUTANE:** Isobutane is a simple asphyxiant (SA). Oxygen levels should be maintained above 19.5%.

**ETHANE:** Guinea pigs breathing about 2.2 to 5% Ethane for 2 hours showed signs of irregular breathing and slight drowsiness, but no other health effects. At concentrations of 15-19%, when mixed with oxygen, Ethane is a weak cardiac sensitizer. There were no signs of anesthesia in animals breathing an ethane/oxygen mixture (80% ethane/20% oxygen) for up to 3.75 hours.

**PROPANE:**

**Skin Contact (Rabbit):** Several formulations containing an isobutane-propane mixture were tested for skin irritation effects. All formulations contained less than 13% propane. All of the formulations containing propane caused only mild irritation. Effects on Short-Term Inhalation: Guinea-pigs breathing 5.5% propane by volume developed tremors after 5 minutes. Nausea, retching, and stupefaction were observed when animals were exposed for 30-120 minutes. All the animals survived a two-hour exposure and had no significant tissue damage. A gas concentration of 89% did not cause anesthesia, but depressed the blood pressure of cats.

**PROPANE (Continued):**

Inhalation of 10 percent propane by mice and 15% by dogs caused weak cardiac sensitization. Presumably, all of these effects are reversible when exposure ceases. In primates, 10% propane caused some change in heart function. At 20% there was aggravation of these symptoms and respiratory depression. Effects of Long-Term Inhalation: No toxicity or abnormalities were observed when monkeys were exposed to approximately 750 ppm for 90 days. Similar results were obtained when monkeys were exposed to an aerosol spray containing 65% propane and isobutane.

**SUSPECTED CANCER AGENT:** None of the components of this gas mixture are found on the following lists: FEDERAL OSHA Z LIST, NTP, CAL/OSHA, IARC, and are therefore not considered to be, nor suspected to be, cancer-causing agents by these agencies.

**IRRITANCY OF PRODUCT:** Contact with rapidly expanding gases can cause frostbite and damage to exposed skin and eyes.

**SENSITIZATION OF PRODUCT:** Ethane, Propane, and n-Butane (components of this product) are not known to cause sensitization in humans; however, some animal studies indicate that exposure to these gases can cause weak cardiac sensitization.

**REPRODUCTIVE TOXICITY INFORMATION:** Listed below is information concerning the effects of this product and its components on the human reproductive system.

**Mutagenicity:** This product is not reported to cause mutagenic effects in humans.

**Embryotoxicity:** This product is not reported to cause embryotoxic effects in humans.

**Teratogenicity:** This product is not reported to cause teratogenic effects in humans.

**Reproductive Toxicity:** This product is not reported to cause adverse reproductive effects in humans.

*A **mutagen** is a chemical which causes permanent changes to genetic material (DNA) such that the changes will propagate through generation lines. An **embryotoxin** is a chemical which causes damage to a developing embryo (i.e. within the first eight weeks of pregnancy in humans), but the damage does not propagate across generational lines. A **teratogen** is a chemical which causes damage to a developing fetus, but the damage does not propagate across generational lines. A **reproductive toxin** is any substance which interferes in any way with the reproductive process.*

**MEDICAL CONDITIONS AGGRAVATED BY EXPOSURE:** Acute or chronic respiratory conditions may be aggravated by over-exposure to the components of this product.

**RECOMMENDATIONS TO PHYSICIANS:** Administer oxygen, if necessary; treat symptoms eliminate exposure.

**BIOLOGICAL EXPOSURE INDICES:** Currently, there are no Biological Exposure Indices (BEIs) associated with the components of this product.

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### **12. ECOLOGICAL INFORMATION**

**ENVIRONMENTAL STABILITY:** This gas will be dissipated rapidly in well-ventilated areas. The components of the gas mixture are stable in the environment. Additional environmental data for the components of this product is available as follows:

**BUTANE:** Log K<sub>ow</sub> = 2.89. Water Solubility = 6.4 ppm at 25°C. Log BCF (n-butane) = calculated, 1.78 and 1.97, respectively. Expected Half-life = 0.13 hr. Bioconcentration factors do not indicate that bioconcentration in aquatic organisms is important.

**PROPANE:** Log K<sub>ow</sub> = 2.36. Water Solubility = 2.62.4 ppm at 25°C. Log BCF = calculated, 1.56 and 1.78, respectively. The bioconcentration in aquatic organisms is not expected to be important.

EFFECT OF MATERIAL ON PLANTS or ANIMALS: Any adverse effect on animals would be related to oxygen deficient environments, as well as respiratory system damage. Frost-produced in the presence of rapidly expanding gases may adversely affect plant life.

EFFECT OF CHEMICAL ON AQUATIC LIFE: No evidence is currently available on this product's effects on aquatic life. However, all work practices should be aimed at eliminating contamination of aquatic environments with this product.

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### **13. DISPOSAL CONSIDERATIONS**

PREPARING WASTES FOR DISPOSAL: Waste disposal must be in accordance with appropriate Federal, State, and local regulations. Return cylinders with residual product to MESA. Do not dispose of locally.

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### **14. TRANSPORTATION INFORMATION**

THIS MATERIAL IS HAZARDOUS AS DEFINED BY 49 CFR 172.101 BY THE U.S. DEPARTMENT OF TRANSPORTATION.

PROPER SHIPPING NAME: Liquefied gas, flammable n.o.s. (Propane, Butane)

HAZARD CLASS NUMBER and DESCRIPTION: 2.1 (Flammable Gas)

UN IDENTIFICATION NUMBER: UN 3161

PACKING GROUP: Not applicable

DOT LABEL(S) REQUIRED: Flammable Gas

NORTH AMERICAN EMERGENCY RESPONSE GUIDEBOOK NUMBER (1996): 115

MARINE POLLUTANT: The components of this gas mixture are not classified by the DOT as Marine Pollutants (as defined by 49 CFR 172.101, Appendix B).

TRANSPORT CANADA, TRANSPORTATION OF DANGEROUS GOODS REGULATIONS: THIS MATERIAL IS CONSIDERED AS DANGEROUS GOODS. Use the above information for the preparation of Canadian Shipments.

## 15. REGULATORY INFORMATION

**SARA REPORTING REQUIREMENTS:** The components of this gas mixture are subject to the reporting requirements of Sections 302, 304 and 313 of Title III of the Superfund Amendments and Reauthorization Act., as follows:

COMPONENT	SARA 302	SARA 304	SARA 313
Butane	NO	NO	NO
Ethane	NO	NO	YES
Iso-Butane	NO	NO	NO
Propane	NO	NO	YES

**SARA Threshold Planning Quantity:** Not applicable.

**TSCA INVENTORY STATUS:** The components of this gas mixture are listed on the TSCA Inventory.

**CERCLA REPORTABLE QUANTITY (RQ):** Not applicable.

**OTHER FEDERAL REGULATIONS:** Methane, Ethane, Isobutane and Propane are subject to the reporting requirements of Section 112(r) of the Clean Air Act. The Threshold Quantity for each of these gases is 10,000 pounds. Depending on specific operations involving the use of this product, the regulations of the Process Safety Management of Highly Hazardous Chemicals may be applicable (29 CFR 1910.119). This regulation is pertinent when the total amount of flammable gas on site is above 10,000 lbs.

**STATE REGULATORY INFORMATION:** The components of this gas mixture are covered under specific State regulations, as denoted below:

**Alaska - Designated Toxic and Hazardous Substances:** Methane, Propane, Butane, Ethane.

**California - Permissible Exposure Limits for Chemical Contaminants:** Methane, Propane, Butane, Ethane.

**Florida - Substance List:** None.

**Illinois - Toxic Substance List:** Methane, Butane, Ethane.

**Kansas - Section 302/313 List:** Ethane.

**Massachusetts - Substance List:** Methane, Propane, Butane, Isobutane.

**Minnesota - List of Hazardous Substances:** Methane, Propane, Butane, Ethane.

**Missouri - Employer Information/Toxic Substance List:** Methane, Propane, Butane, Ethane.

**New Jersey - Right to Know Hazardous Substance List:** Methane, Propane, Butane, Ethane, Isobutane.

**North Dakota - List of Hazardous Chemicals, Reportable Quantities:** None.

**Pennsylvania - Hazardous Substance List:** Methane, Propane, Butane, Ethane 2-Methyl-Propane.

**Rhode Island - Hazardous Substance List:** Methane, Propane, Butane, Ethane.

**Texas - Hazardous Substance List:** Propane.

**West Virginia - Hazardous Substance List:** Propane.

**Wisconsin - Toxic and Hazardous Substances:** Propane.

**CALIFORNIA PROPOSITION 65:** No component of this product is on the California Proposition 65 lists.

**LABELING (For Compressed Gas):**

**DANGER:**

FLAMMABLE LIQUID AND GAS UNDER PRESSURE.  
CAN FORM EXPLOSIVE MIXTURES WITH AIR.  
MAY CAUSE FROSTBITE.

Keep away from heat, flames, and sparks.  
Store and use with adequate ventilation.  
Cylinder temperature should not exceed 125 °F (52 °C).  
Do not get liquid in eyes, on skin or clothing.  
Close valve after each use and when empty.  
Use in accordance with the Material Safety Data Sheet.

**FIRST-AID:**

**IF INHALED,** remove to fresh air. If not breathing, give artificial respiration. If breathing is difficult, give oxygen. Call a physician.

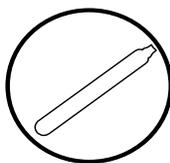
**DO NOT REMOVE THIS PRODUCT LABEL.**

**TARGET ORGANS:** Respiratory system.

**WHMIS SYMBOLS:**

**A:** Compressed Gas

**B-1:** Flammable Gas



## DEFINITIONS OF TERMS

A large number of abbreviations and acronyms appear on a MSDS. Some of these which are commonly used include the following:

**CAS #:** This is the Chemical Abstract Service Number which uniquely identifies each constituent. It is used for computer-related searching.

### EXPOSURE LIMITS IN AIR:

**ACGIH** - American Conference of Governmental Industrial Hygienists, a professional association which establishes exposure limits.

**TLV - Threshold Limit Value** - an airborne concentration of a substance which represents conditions under which it is generally believed that nearly all workers may be repeatedly exposed without adverse effect. The duration must be considered, including the 8-hour **Time Weighted Average (TWA)**, the 15-minute **Short Term Exposure Limit**, and the instantaneous **Ceiling Level**. Skin adsorption effects must also be considered.

**OSHA** - U.S. Occupational Safety and Health Administration.

**PEL - Permissible Exposure Limit** - this exposure value means exactly the same as a TLV, except that it is enforceable by OSHA. The OSHA Permissible Exposure Limits are based in the 1989 PELs and the June, 1993 Air Contaminants Rule (Federal Register: 58: 35338-35351 and 58: 40191). Both the current PELs and the vacated PELs are indicated. The phrase, "Vacated 1989 PEL", is placed next to the PEL which was vacated by Court Order.

**IDLH - Immediately Dangerous to Life and Health** level represents a concentration from which one can escape within 30-minutes without suffering escape-preventing or permanent injury.

**The DFG - MAK** is the Republic of Germany's Maximum Exposure Level, similar to the U.S. PEL. **NIOSH** is the National Institute of Occupational Safety and Health, which is the research arm of the U.S. Occupational Safety and Health Administration (**OSHA**). NIOSH issues exposure guidelines called **Recommended Exposure Levels (RELs)**. When no exposure guidelines are established, an entry of **NE** is made for reference.

### FLAMMABILITY LIMITS IN AIR:

Much of the information related to fire and explosion is derived from the **National Fire Protection Association (NFPA)**. **L<sub>EL</sub>** - the lowest percent of vapor in air, by volume, that will explode or ignite in the presence of an ignition source. **U<sub>EL</sub>** - the highest percent of vapor in air, by volume, that will explode or ignite in the presence of an ignition source.

### TOXICOLOGICAL INFORMATION:

Possible health hazards as derived from human data, animal studies, or from the results of studies with similar compounds are presented. Definitions of some terms used in this section are: **LD<sub>50</sub>** - Lethal Dose (solids & liquids) which kills 50% of the exposed animals; **LC<sub>50</sub>** - Lethal Concentration (gases) which kills 50% of the exposed animals; **ppm** concentration expressed in parts of material per million parts of air or water; **mg/m<sup>3</sup>** concentration expressed in weight of substance per volume of air; **mg/kg** quantity of material, by weight, administered to a test subject, based on their body weight in kg. Data from several sources are used to evaluate the cancer-causing potential of the material. The sources are: **IARC** - the International Agency for Research on Cancer; **NTP** - the National Toxicology Program, **RTECS** - the Registry of Toxic Effects of Chemical Substances, **OSHA** and **CAL/OSHA**. IARC and NTP rate chemicals on a scale of decreasing potential to cause human cancer with rankings from 1 to 4. Subrankings (2A, 2B, etc.) are also used. Other measures of toxicity include **TDL<sub>0</sub>**, the lowest dose to cause a symptom and **TCL<sub>0</sub>** the lowest concentration to cause a symptom; **TDo**, **LDLo**, and **LDo**, or **TC**, **TC<sub>0</sub>**, **LCL<sub>0</sub>**, and **LC<sub>0</sub>**, the lowest dose (or concentration) to cause death. **BEI** - Biological Exposure Indices, represent the levels of determinants which are most likely to be observed in specimens collected from a healthy worker who has been exposed to chemicals to the same extent as a worker with inhalation exposure to the TLV.

### REGULATORY INFORMATION:

This section explains the impact of various laws and regulations on the material. **EPA** is the U.S. Environmental Protection Agency. **WHMIS** is the Canadian Workplace Hazardous Materials Information System. **DOT** and **TC** are the U.S. Department of Transportation and Transport Canada, respectively. These are: **Superfund Amendments and Reauthorization Act (SARA)**; the **Toxic Substance Control Act (TSCA)**; Marine Pollutant status according to the **DOT**; California's Safe Drinking Water Act (**Proposition 65**); the **Comprehensive Environmental Response, Compensation, and Liability Act (CERCLA or Superfund)**; and various state regulations. This section also includes information on the precautionary warnings which appear on the materials package label.

The information contained herein is based on data considered accurate. However, no warranty is expressed or implied regarding the accuracy of these data or the results to be obtained from the use thereof. ComStar assumes no responsibility for injury to the vendee or third persons proximately caused by the material if reasonable safety procedures are not adhered to as stipulated in the data sheet. Additionally, ComStar assumes no responsibility for injury to vendee or third persons proximately caused by abnormal use of the material even if reasonable safety procedures are followed. Furthermore, vendee assumes the risk in his use of the material.